Atoms CBRT Script By Sasha Bratt

Atoms!

1: Ladies and gentlemen!

2: Boys and girls!

3: Children of all ages!

4: Introducing...[Sound effect-Drum

1-4: The Atom!!! [Gesture] [Sound effectl

5: Where is it?

19: There. [Gesture]

6: I don't see it.

1-4: There! [Gesture]

20: Atoms are everywhere!

7-8: Why can't we see them?

21: They are microscopic. [Gesture]

9: Microscopic?

10: That means small.

11-15: Duh!

22: Atoms are the building blocks of life.

16: Like Legos?

17: Or Lincoln logs?

1-2: Yes.

3-4: ...and no.

5-18: [Sound effect – Confused]

18: So what is an atom?

19: An atom is basic unit of matter.

6: What's matter?

2: Nothin'...what's a matter with you?

All: [Sound effect-Laughter]

3: Matter is anything that takes up space.

4: And all matter is made up of atoms!

5-18: Even us? [Gesture]

1-4: Even you! [Gesture]

7: So, what is in an atom?

20: Good question!

21: An atom has three main

components.

22: Protons!

1: Neutron!

4: And electrons!

2-3: Oh my! [Gesture]

8-10: Protons, neutron, electrons...

11-18: Oh my! [Gesture]

3: The protons, neutrons and electrons

all have electrical charges.

All: [Sound effect] [Gesture]

11: What are their electrical charges?

4: Great question!

11: Thanks.

19: Protons are positive. [Gesture]

20: Neutrons are neutral. [Gesture]

12: What does that mean?

21: It means that their charge is neither positive nor negative.

22: And electrons are negative.

[Gesture]

1: The protons and neutrons make up the middle of the atom. It's called the nucleus.

13: Gross...mucus! [Sound effect]

All, except 13: Not mucus...nucleus!

13: Oh...sorry.

14: Where do the electrons go?

20: They surround the nucleus in a cloud.

15: Why don't they float away?

22: Because they are bound to the nucleus by an electromagnetic force!

All: WHOA!

16: That sounds like science fiction! [Sound effect]

1-2: No, its just science!

7: So my shirt it made of atoms?

1-4: Yes!

8: And my lunch?

19-22: YES!

17: and everything else? [Gesture]

22: Yes, atoms are everywhere, all the time.

5-18: Awesome!

1: And if you forget what's in an atom, just remember...

1-7: Protons,

8-15: Neutrons.

1-22: Electrons,

All: OH MY!!!!!!!!